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#### B.Sc Part-I

# Biolechnology

Main Subject: Biotechnology (Summary of the Course Structure)
Compulsory Subjects: Behavioural Science and Entrepreneurship
Development

#### B.Se. - I (At the end of First Year)

	Examinations	Total Marks
1.	Four Theory paper (I - IV) of 75 marks each	300
2.	Practical based on papers I - IV	100
3.	Behavioural Science	50
4.	Internal Assessment	50
On-	the-job-training (30 days-duration summer) Report	
	B.Sc II (At the end of Second Years)	
5.	Four Theory papers (V-VIII) of 75 marks each	300
6.	Practical based on papers V - VIII	100
7.	Internal Assessment	100
On-	the-job-training (30 days-duration summer) Report	
	B. Sc III (At the end of Third Year)	
8.	Five Theory papers (IX-XIII) of 75 marks each.	375
9.	Practical based on papers IX - XIII	100
10.		75
11.		150
12:	Internal Assessment	100
	ject work of 6 month duration under supervision of a cho	osen faculty
	Total	1800

Note: 1. Each theory paper shall be of three-hour duration

2. Each practical examination shall be of six hour duration.

3. 20% of the marks in each practical examination shall be reserved for class records.

 Internal assessment shall be based on training report, oral presentation and term paper.

### Paper I: Biochemistry and Biophysics

Nature of Blological materials: Polymeric reaction, carbohydrates, lipids, proteins, nucleotide, nucleic acid, oxidation-reduction properties. pH, pK and buffering, isomerism, types of chemical bonds, and hydrophilic

hydrophobic group n biomolecules, nemotransmatters, hormones and growth factors, high energy biomolecules (ATP, GTP & Creatine phosphate)

Perspectives of biologica! macromolecules: The repeating units in nucleic acids and proteins, helicity, bending, looping pleats, salt bridges, etc. and their determinants, basis for intermolecular interactions with example, enzymesubstrate and antigen-antibody reactions, salient features of biochemical reactions involved in the biosynthesis of amino acids, fatty acids and nucleotides.

Enzymes, Protein and non-protein enzymes: Classification and nomenclature of enzymes, regulation of enzyme activity, coenzymes structure and function of Coenzyme I and Coenzyme A; kinetics of enzyme catalysed reactions; isolation and purification of enzymes; enzymes in food processing, medicines, and production of chemical compounds.

Bio-energetics: Laws of thermodynamics (1st & 2nd laws), electrical properties of biological compartments, electrochemical gradients, membrane potential, chemiosmotic hypothesis.

Energetics of a living body: Sources of heat limit to temperature, heat dissipation and conservation; Lambert-Beer law, spectrophotometry nad collorimetery, primary events in photosynthesis, strategies of high reception in microbes, plants and animals, correction of vision bulb generation and reception of sonic vibrations, hearing aids

Electrical properties of biological compartments: Electricity as a potential signal.

Intra and intermolecular interactions in biological system: Space and charge compatibility as determinant of such interactions.

# Paper II: Maths, Biostatistics, Computers and Bioinformatics

Maths: The set theory properties of sub sets; linear geometric functions, the binomial theorem of integer, limits of functions (basic idea of limits of function without analytic definition), derivatives of functions, logarithm (definition & laws of logarithm, use of logarithm table), differentiation, integration (general introduction, significance and application for simple algebraic and trigonometric functions).

Biostatistics: Probability calculation (classical & axiomatic definition of probability, theorem on total and compound probability), standard

\* Estimation of sugar in given solution.

Cell Biology

\* Cytological preparations

Fixation, dehydration and staining

Biostatistics

\*

Computers

\* Handling of computer and Data analysis

# B.Sc. in Biotechaology Behavioural Science (Compulsory Subject)

## Syllabus for Behevioural Science

50 Marks: 3 Hours

(a) Psychology

- 1. Natur@and Methods of Psychology
- 2. Attention and perception
- 3. Motivation
- 4. Emotion
- 5. Learning and Memory
- 6. Thinking
- 7. Intelligence
- 8. Personality its nature and assessment

#### (b) Anthropology

#### Cultural basis of behaviour-Nature and Natures.

Process of Socialization
Enculturation
Value Orientation
Culture and Society
Nature of Culture and its Development
Society-its characteristics
Types and organization

#### Genetic basis of behaviour-Primate and Human

Evolution of behaviour

Genetic and ueviant behaviour

Mechanism and development of behaviour phenotypes.

## Inorganic and Physical Chemistry

(40 Hours)

Unit-I

Inorganic

Chemical bonds and molecules: Shapes of simple molecules, bond energy, bond length, resonance and Hydrogen bond.

Radioactivity: Natural and artificial, group displacement law, half life period, binding energy, nuclear reaction equations, isotopes, tracers, radiodating, Application of radioactivity.

Periodic table: Modern periodic table, periodicity in properties of element, atomic radii, ionic and covalent radii, ionization energies, electron affinity, electron-negativity.

Ores and Minerals: Principles involved in the extraction of metals from ores, including their refining and purification. General trends in the Chemistry of S-block elements (Group IA & IIA).

Unit-II

(20 Hours)

Physical

Gases: Kinetic theory of gases, van der Waal's equation, critical constants, Liquefaction of gases.

Chemical Kinetics: Velocity of a reaction, Law of mass action; determination of rate constant for first and second order reactions, collision theory of bimolecular reactions.

Catalysis: Promoters and Poisons, Enzyme catalysis.

### Paper II-Organic and Physical

Unit-I

Organic

Organic chemistry is Chemistry of carbon compounds. Methods of purification, tests of purity; qualitative and quantitative elemental analysis, determination of molecular formula. Tetrahedral concept in carbon compounds, functional groups, nomenclature and isomerism.

- \* General methods of preparation and properties of alkane, alkenes, alkynes, halogen substituted alkanes (CH2, Cl2, CHC14, CCl4, ClH2) and others.
- Propanol; glycol, glycerol, aldehydes, ketones, formaldehyde, acetaldehyde, and acetone. Monocarbosylic acids and their simple derivatives, Descriptive studies of dicarboxylic acids, viz malic oxalic, tartaric, malcic, fumaric acits. Tri-carboxylic acid, citric acid and urea stereoisomerism; geometrical and optical.

Keto-enol tautomerism; aceto-acetic ester and malonic ester Petroleum; Fractionation, cracking and synthetic petrol.

#### Unit-II Physical

(20 Hours)

Chemical equilibrium: Reversible reactions, equilibrium law equilibrium constant factors influencing equilibrium states.

Electrochemistry: Electrolysis, laws of electrolysis, ionization specific, equivalent and molecular conductance, ionization constant common ion effect; Hydrogen ion concentration. pH value. Theory of acid base indicators, buffer solutions, hydrolysis of salts and solubility product simple calculations based on these concepts.

#### Practicals

#### Inorganic

- 1. Volumetric analysis: Oxidation-reduction titration using KMnO<sub>4</sub> and K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>.
- 2. Iodometry titrations: Estimation of sodium thiosulphate, potassium dichromate, and copper sulphate.
- Preparation of the following inorganic compounds: Prussion blue from iron fillings, chrome alum, cuprammonium sulphate, cuprous chloride and potassium trioxide atrochromate.

#### Physical

- (1) Determination of surface tension and viscosity of liquid.
- (2) Heat of neutralization of a strong acid and a strong base.
- (3) Solubility curve, KNO3 or benzoic acid.

## B. Sc. in Biotechnology Botany (Subsidiary Subject)

There sh	art-I (At the end of First year) all be Examination in two apers and a practical	Duration (Hours)	Max Marks
Paper I	Algae, Fungi, Bryophytes and Pteridophytes	3	50
Paper II	Gymnospoerms & Taxonomy of Angiosperms	3	50 ,
Practical	examination based paper I & II	4	50

20% of the Marks in the Practical examination shall be reserved for the class and field records.

# B.Sc. Part-II (At the end of Second year)

There shall be Examination in two theory papers and a practical		Duration (Hours)	Max. Marks
Paper III	Plant Physiology, Ecology and Plant Anatomy	3	50
Paper IV	Embryology, Plant Pathology, Plant Breeding & Economic Botan	3 iy	50
Practical	examination based on Paper I and	II 4	50

20% of the Marks in the Practical examination shall be reserved for the class records.

# Paper - I (Algae, Fungi, Bryophytes and Pteridophytes)

Unit-I (15 Hrs)

Classification, occurrence, structure, systematic position of mode of reproduction and economic importance of the following general; Chlamydomonas, Volvox, Oedoponium, Vaucheria, Chara, Saragassum Polysiphonia Nostoc.

Unit-II

(15 Hrs.)

- 1. Outlines of classification of fungi.
- Systematic position, occurrence, streutreamd mode of reproduction in fungi, based on the following representatives: Synchytrium, Eurotium, Morchella, Agaricus and Alternaria.
- 3. Economic importance of fungi.
- 4. Lichens: Classification, occurrence, systematic position, mode of nutrition, reproduction and economic importance of Lichens.

Unit-III

(15 Hrs.)

- 1. Outlines of classification and importance of bryophytes.
- 2. Systematic position occurrence, morphology, anatomy and reproductive structure in Riccia, Marchantia, Anthocerus (Development of Sporophyte only)

Unit-IV

(15 Hrs.)

- Systematic position, occurrence, morphology, anatomy and development of reproductives of the following taxa-Rhynia, Selagnella, Equisetum and Marsilca.
- 2. Steler system and its evolution in Pteridophytes.
- 3. Heterospory and seed habit.

#### Practicals

Algae Study of Algal types with the help of permanent slides and also by preparation suitable slide as presented in the theory course.

Fungi Study of Fungal types with the helped permanent slides and also by preparing suitable slide and also by preparing suitable slide as prescribed as the theory course.

Study of the following types with the help of permanent slides and also by cutting sections and making suitable preparations.

Bryophytes :

Riccia, Marchantia, Anthoceros.

Pteridophytes:

Rhymia, Selanginella, Equisetum and Marsilea

#### Practicals

Paper - 2

(Gymnosperms & Taxonomy of Angiosperms)

Unit - 1

(20 Hrs.)

- 1. General characteristics, affinities and classification of Gymnosperms.
- 2. Distribution of Gymnosperms in India.
- 3. Systematic position, occurrence, morphology and development, of reproductive structures of the following taxa-Cycas, Pinus, Ephedra
- 4. Economic importance of Gymnosperns.

Unit - 2

(10 Hrs.)

- 1. Classification as proposed by Bentham and Hooker and Hutchison.
- 2. Important Herbaria and Botanical gardens of India.
- 3. Binomial Nomchelature and elementary knowledge of international Code of Botanical Nomenclature.

Unit - 3

(30Hrs.)

- 1. Systematic position, distinguishing character and economic importance of the following families.
- 2. Dicotyledons:
  - (a) Polypetatae: Ranunculaceae, papaveraceae, Caryophyllaceae, Rutaceae, Cucurbitaceae, Rosaceae, Apiaceae.
  - (b) Gamopetalae, Rubiaceae, Apocynaceae, Asclepiadaceae, Convolvulaceae, Acanthaceae, Lamiaceae, Verbenaceae.
  - (c) Monochlamydeae Euphorbiaceae.
- 3. Monocotyledons: Arecaceae, Poaceae.

#### PRACTICALS

Study of the following types with the help of permanent slides and also by cutting sections and making suitable preparations

Gymnosperms:

Cyacas, Pinus, and Ephedra

Taxonemy

Detailed description and identification of locally

available plants of the families as prescribed in theory

course.

distribution with important properties, simple problems involving binomial, poissonand normal variables, methods of sampling collection of data; primary & secondary data, classification & tabulation, confidence level, statistics, idea of sampling, distribution and standard error, large samples; normal tests, measurement of dispersion (measures of location and dispersion).

Computers: General introduction (characteristics, capabilities, generations), software, hardware; organic ion of hardware (input devices memory, control unit arithmetic logic unit, output devices); software; application software, languages-low level, high level). interpreter, compiler, data processing; batch, on-line, real time (example from bio-industries, e.g. application of computers in co-ordination of solute concentration, ph, temperature, etc., of a fermenter in operation); internet application.

Bioinformatics: Application of computers in biotechnology, genome analysis, sequence, primer designing, phylogenetic analysis.

### Paper III: Cell Biology

Cell as a basic unit of living systems: The cell theory precellular evolution; artificial creation of "cell"; broad classification cell type. PPLOS, bacteria, eukaryotic microbes, plant-and animal cells a details classification cell types within an organism, cell, tissue, organ and organism at different levels of organisation of otherwise genetically similar cells, ecological amplitude of cells in high altitude, sediments, areties, hotspring, arid, brackish and freshwater environments; biochemical composition of cells (proteins, lipids, carbohydrates, nucleic acids and the metabolic pool.

Ultra structure of the cell membrane and cell organelles: Structure and function of cell organelles; ultrastructure of cell membrane, cytocol golgi bodies, endoplasmic eticeulum (rough and smooth), mitochondria. Chloroplasts, lysosomes, peroxysomes, nucleus (nuclear membrane nucleoplasm, nucleolus).

Chromosomes: Chemical composition; structural organic atoms of chromatids, centromeres, telomeres, chromatin, nucleosome organisation eu-and heterochromatin; special chromosomes (e.g., polytene and lampbrush chromosomes); banding patterns in human chromosomes.

Cell division and cell cycle: mitosis and meiosis interphase and mitosis comparisin of mitosis and meiosis

Cell-Cell interaction. Cell locomotion (amoeboid, flagellar, and cililar) muscle and nerve cells, cell senescence and death.

Cell differentiation: Plants and animals difference between normal and cancer cells.

# Paper IV: Instrumentation & Bio-analytical Techniques

Microscopy: Simple microscopy, phase contrast microscopy, florescence and electron microscopy (TEM and SEM)

Instruments, basis principles and usage: pH meter, absorption and emission spectroscopy, principle and law of absorption and radiation, use of densitometry, fluorimetry, colorimetry, spectrophotometry (visible, uv, infrared), manometry, polarography, centrifugation, atomic absorption, IR, NMR, X-ray crystallography.

Chromatography techniques: Paper chromatography, thin layer chromatography, colomn chromatography, gas chromatography, affinity chromatography, gel filtration, chromotography, ion exchange (paper gel etc.)

Electrophoresis: SDS polyacrylamide electrophoresis, immunoelectrophoriss, isoelectric focussing.

Fermentaton: Different type of fermenters: principle operating, characteristics of fermenters, computer control of fermentation process.

Radioisotope tracer techniques and autoradiography.

#### **Practicals**

## **Practical Examination**

Biochemistry

- \* Preparation of buffer
- Colour reactions of Carolydrates
- \* Colour reactions of amino acids
- \* Extraction of lipids
- Estimation of protein by Bradford method
- \* Estimation of DNA
- \* Estimation of RNA